

# How to determine empirical factors

- $k_f$  (gas film mass transfer coefficient)

- based on diluted packed bet experimental

$$Sh : Sh = 1.42 + 0.92 Sc^{1/3} Re^{0.6}$$

- $D_{eff}$  (effective diffusion coefficient)

- based on Knudsen and molecular diffusion

- approximated by Modified Bruggeman<sup>[1]</sup>

$$D_{eff} : D_{eff} = \frac{1}{\frac{1}{D_{pore}} + \frac{1}{D_{CO_2, N_2}}} \quad \text{constant}$$

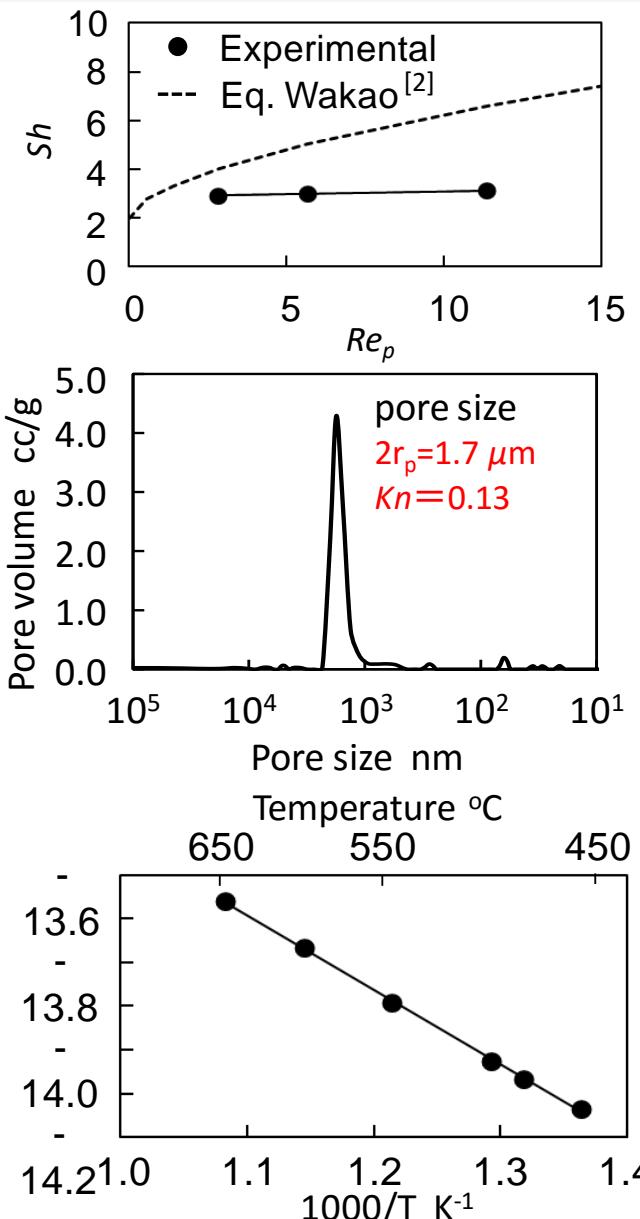
$$D_{Kn} = \frac{4}{3} r_p \sqrt{\frac{2RT}{\pi M_{CO_2}}} \quad D_{CO_2, N_2} = \frac{0.00266 T^{3/2}}{PM_{CO_2, N_2}^{1/2} \sigma_{CO_2, N_2}^2 \Omega_D}$$

- $k_c$  (surface chemical reaction coefficient )

- based on experimental data

of 100% CO<sub>2</sub> conditions

$$k_c : k_c = 1.16 \times 10^{-1} \exp\left(-\frac{1.42 \times 10^4}{RT}\right)$$



[1] D.A.G. Bruggeman ;Ann. Physik. 24 636 1935

[2] N. Wakao and T. Funazkri : chem Eng. sci 33 1375(1978)